

Competition Corner: Electron Accounting

g11_u05_w18

Original fictional paper-model challenges. No apparatus or procedure is implied. Use $F = 96\,485\text{ C mol}^{-1}\text{ e}^{-}$, $M_{\text{Cu}} = 63.55\text{ g mol}^{-1}$, and $M_{\text{Al}} = 26.98\text{ g mol}^{-1}$.

Challenge 21. A fictional allocation card supplies $Q = 5789.1\text{ C}$. Exactly 60.0% of the modeled electron amount is assigned to $\text{Cu}^{2+} + 2\text{e}^{-} \rightarrow \text{Cu}$; the remaining 40.0% is assigned to $\text{Al}^{3+} + 3\text{e}^{-} \rightarrow \text{Al}$. Find the modeled mass of each product and the total modeled mass. State why this allocation is a bookkeeping assumption, not evidence of a mechanism.

Challenge 22. A paper data set reports $I = 0.800 \pm 0.010\text{ A}$, $t = 250. \pm 2. \text{ s}$, and a modeled copper mass calculated from their product. Estimate the largest relative uncertainty in $Q = It$ by adding the two relative uncertainties. Then name one measurement not supplied that would still prevent an efficiency claim from this calculation alone.